Applicant: William L. Bowden et al. Attorney's Docket No.: 08935-251002 / M-4971A

Serial No.: 10/796,739 Filed: March 9, 2004

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Amendments to the Claims:

This listing of claims replaces all prior versions and listings of claims in the application:

<u>Listing of Claims</u>:

1-23. (Cancelled)

24. (Currently amended) A method of manufacturing an electrochemical cell, the method comprising:

providing an a positive electrode including a lambda-manganese oxide; and

after providing the positive electrode, forming a cell including the positive electrode and
a lithium negative electrode,

wherein the cell has a closed circuit voltage of about 4V and a specific discharge capacity at a nominal discharge rate of 1 mA/cm² to a 3V cutoff of greater than 120 mAh/g.

25. (Currently amended) The method of claim 24, wherein providing the <u>positive</u> electrode includes preparing lambda-manganese dioxide by a method comprising:

contacting water with a compound of the formula $\text{Li}_{1+x}\text{Mn}_{2-x}\text{O}_4$, wherein x is from -0.02 to +0.02;

adding an acid to the water and compound until the water has a pH of 1 or less; separating a solid from the water and acid; and

drying the solid at a temperature of 120°C or below to obtain the lambda-manganese dioxide.

26. (New) The method of claim 25, wherein the compound has a BET surface area of between 1 and $10 \text{ m}^2/\text{g}$.

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27. (New) The method of claim 25, wherein the compound has a spinel-type crystal structure.

- 28. (New) The method of claim 25, wherein the solid is dried at a temperature between 30°C and 90°C.
- 29. (New) The method of claim 25, wherein the solid is dried at a temperature between 50°C and 70°C.
- 30. (New) The method of claim 25, wherein x is from -0.005 to +0.005.
- 31. (New) The method of claim 25, wherein contacting water and the compound includes forming a slurry.
- 32. (New) The method of claim 31, wherein the slurry is maintained at a temperature below 50°C.
- 33. (New) The method of claim 31, wherein the temperature of the slurry is held substantially constant during the addition of acid.
- 34. (New) The method of claim 25, wherein the acid comprises sulfuric acid, nitric acid, perchloric acid, hydrochloric acid, toluenesulfonic acid or trifluoromethylsulfonic acid.
- 35. (New) The method of claim 25, wherein the pH is 0.7 or less.
- 36. (New) The method of claim 25, wherein the acid has a concentration of between 1 and 8 molar.
- 37. (New) The method of claim 25, further comprising washing the solid separated from the liquid phase with water until the washings have a pH of between 6 and 7.

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38. (New) The method of claim 24, wherein the cell comprises a primary cell.

- 39. (New) The method of claim 24, wherein the cell has a specific discharge capacity at a nominal discharge rate of 1 mA/cm² to a 3V cutoff of greater than 130 mAh/g.
- 40. (New) The method of claim 24, wherein the cell has a specific discharge capacity at a nominal discharge rate of 1 mA/cm² to a 3V cutoff of greater than 135 mAh/g.
- 41. (New) The method of claim 24, wherein the cell has a specific discharge capacity at a nominal discharge rate of 1 mA/cm² to a 3V cutoff of 140 mAh/g or greater.
- 42. (New) The method of claim 24, wherein the lambda-manganese oxide has a BET surface area of greater than $4 \text{ m}^2/\text{g}$.
- 43. (New) The method of claim 24, wherein the lambda-manganese oxide has a BET surface area of greater than $8 \text{ m}^2/\text{g}$.
- 44. (New) The method of claim 24, wherein the lambda-manganese oxide has a total pore volume of from 0.05 to 0.15 cubic centimeters per gram.